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covalent_ 1. A group of atoms united by ionic bonds is called a molecule.
_____T_____ 2. A covalent bond is formed by a shared pair of ...

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(continued) Write Lewis structures for
each of the following molecules. Indicated
the bonds with either dots of dashes. 15.
NH₃ 18. PCl₃. 16. H₂ 19. CCl₄. 17. C₂H₄
20. C₂H₆. Answer the following questions

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as directed. 21. Explain why the molecules SF₄ is an exception to the octet rule.\

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S sunlevel 5-2 Review and Reinforcement (continued) Identify each of the following elements as a metal (M), nonmetal (NM), or semimetal (SM). M 19. sodium SM 20. silicon NM 21. neon M 22. calcium NM 23. nitrogen Write the family names that have been given to each of the following groups.

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Wave Properties Review from Glencoe Reinforcement S8P4f. Diagram the parts of the wave and explain how the parts are affected by changes in amplitude and pitch
2. How do changes in one part of a wave affect other parts of a wave? TEXTBOOK PAGES 466-470 *Essential Vocabulary listed in the Standards **Supplemental Vocabulary listed in the

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