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The periodic table and atomic theory Use the following Bohr model to answer questions 1 to 6. Match the Term on the left with the corresponding Number on the right. Each Number may be used more than once. Refer to the diagram above.

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Bohr Model Practice For each element, write the total number of electrons on the line. Then color the correct number of electrons for each orbit. Remember, fill the orbit closest to the nucleus first, but never exceed the number each orbit can hold.

Bohr Model Worksheet - St. Francis Preparatory School

ASSIGNMENT #1: Bohr Model Practice, Worksheet pages 6-8 This assignment is to be completed below in the space provided. Use the innermost circle as the nucleus, and fill the electron shells with the correct number of electrons for each of the first 20 elements in the Periodic Table. eg. Hydrogen has been completed for you as an example. 1.

Unit 2 Chemistry

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