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included in the Lewis Structure. $4 + 3(6) + 2 = 24$ (carbon has four and each oxygen has six; add two for the -2 charge) $5 + 3(1) = 8$ (nitrogen has five; each hydrogen has one) Step #2: Draw the " skeleton structure " with the central atoms and the other atoms, each connected with a single bond. Step #3: Add six more electron dots to each atom except the

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Worksheet [EBOOK] Chemquest 23 Lewis Structures Answer Key [EBOOK] chemquest 24 more lewis structures 24 ChemQuest 29 Monc LcW s Structunes tr I Name Date: Hour: Introduction Questions Draw the Lewis structures for the following molecules. Part c is done for you. a) H₂O (oxygen is central) b) SO₂ c) Nz It

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