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~~configurations~~

~~and what they~~

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~~Diagrams Made~~

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~~Draw Lewis Dot~~

~~Structures~~

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for class 10*

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27 Questions

Show answers.

Question 1 .

SURVEY . 30

seconds . Q.

What atom
matches this
electron

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Configuration?

1s 2 2s 2 2p 6

3s 2. answer

choices . Neon.

Magnesium.

Aluminum. ...

The electron

configuration of

an atom is 1s 2

2s 2 2p 6. The

number of

valence

electrons in the

atom is . answer

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choices . 3 . 6 .

8 . 10 . Tags :

Question 14 .

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What atom

matches this

electron configu

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Configuration? $1s^2 2s^2 2p^6$

$3s^2 3p^6 4s^2 3d^{10}$

... 33 Questions

Show answers.

Question 1 .

SURVEY . 30

seconds . Q. How

many electrons

can an p

sublevel hold?

... What

electron

configuration

matches an

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oxygen atom?

answer choices .

1s 2 2s 2 2p 6

3s², 3p 6 4s 2

3d 10 4p 5. 1s 2

2s 2 2p 4.

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otes-Name: _____P

eriod: ____ Intro-

Answer the

following

questions

concerning

Potassium (K) .1.

How many valence

electrons does

it have? 2. Do a

sketch of its

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atom (Bohr

model) showing
the electron
shells) Quantum

Mechanical Model

of the Atom-The

electron does

NOT have an

exact orbit

around the

nucleus (like

bohr's model)

Atomic orbital-

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Solution for
Consider the
electron
configuration
 $1s^2 2s^2 2p^6$
 $3s^2 3p^6 3d^8 4s^2$.

Which of the
following
statements is
TRUE? (A) There
are 14 electrons
on the third...

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and fourth
periods.

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of the 3d
transition
metals.

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currently
selected item.
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and
diamagnetism.

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principle.

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electrons.

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electrons and

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Read; What is the electron configuration for chromium?

K.Paul Answered:

Dec 11, 2017.

The correct answer to the question "What is the electron configuration for chromium?" is letter B or $1s^2 2s^2 2p^6 3s^2$

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3p⁶ 4s¹ 3d⁵. The actual electron configuration of Cr is [Ar]

4s¹...

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Questions. For
each of the
following
questions,
choose the
letter that best
answers the
question. Show
all questions <=
=> How many
electrons are
there in the

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third shell
(principal
energy level)
Answers

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To answer the
question, you

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need to know the electron configuration of the atoms to see which one has 7 unbonded electrons, like chlorine.

Fluorine, element number 9, has 2 electrons in the s sublevel (K shell). The L

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shell is
incompletely
filled, with 7
electrons. 9.

*Atomic Structure
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38. The ground
state electron
configuration of
Fe is _____ . A.

2 1s 2s 2 3s 3p6
3d6 B. 2 1s 2 2s

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2p⁶ 3s² 3p⁶ 4s

3d⁶ C. 2 1s² 2s

2p⁶ 3s² 3p⁶ 4s²

D. 2 1s² 2s 2p⁶

3s 3p⁶ 4s² 4d⁶

39. The ground state electron configuration of Ga is _____ . A.

2 1s 2s² 3s² 3p⁶

3d¹⁰ 4s² 4p¹ B.

6 1s 2s 2p 3s²

3p⁶ 4s² 4d¹⁰ 4p¹

C. 2 1s² 2s 2p⁶

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solutions. Write
out the orbital
box diagram and
the...

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Which of the following is the correct electron configuration for the bromide ion, Br^- ? a)

[Ar] 4s 2 4p 5.

b) [Ar] 4s 2 3d 10 4p 5. c) [Ar] 4s 2 3d 10 4p 6.

d) [Ar] 4s 2 3d 10 4p 6 5s 1. e)

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[Ar] 4s 2 3d 10

3p 6. answer .

text

explanation.

6-89 . Which of the following elements has the largest number of electrons for which the principal quantum ...

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Answers